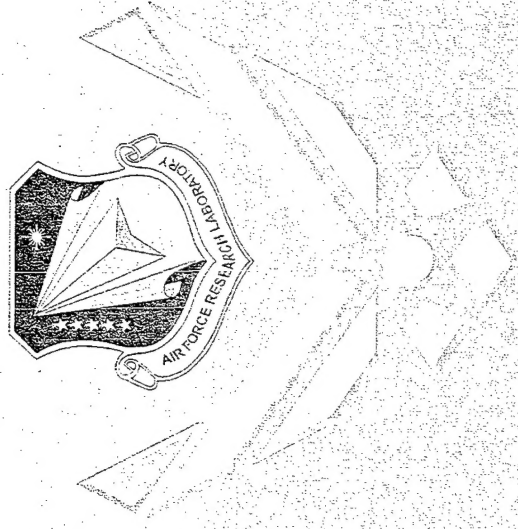


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# The Specific Refractive Index Increments for POSS Polymers in Solution.

Sherly R. Largo<sup>‡</sup>, Timothy Haddad<sup>‡</sup>, Constance  
Schlaefer\* and Rene Gonzalez\*

<sup>‡</sup>ERC, Inc. – \*Air Force Research Laboratory

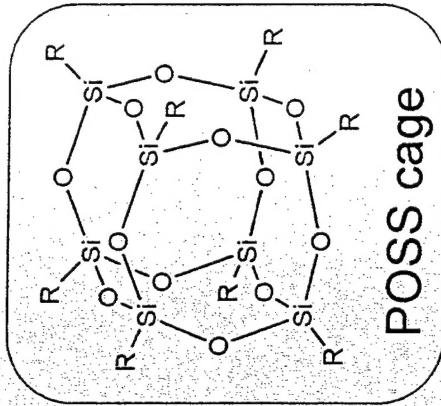
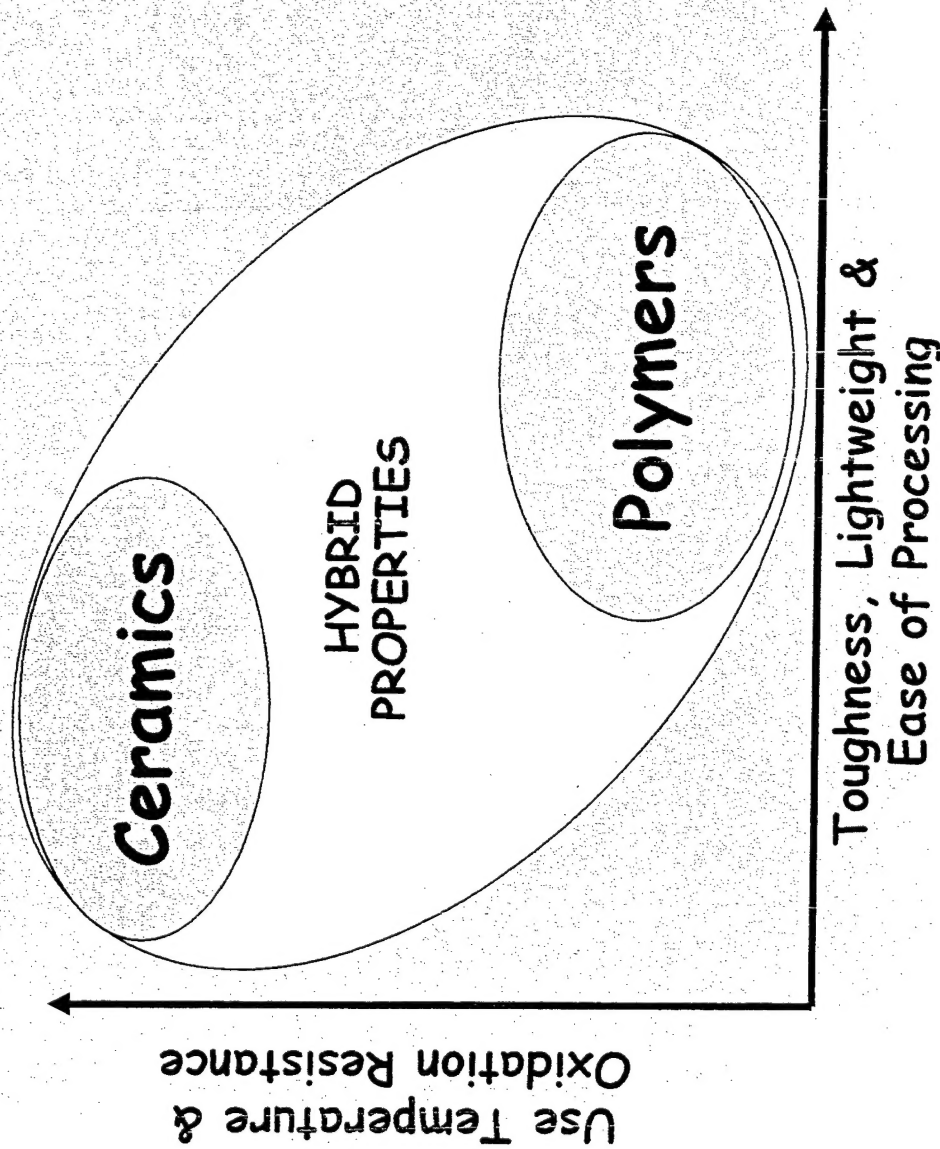
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# Hybrid Inorganic/Organic Polymers

Goal: Develop High Performance Polymers that REDEFINE material properties

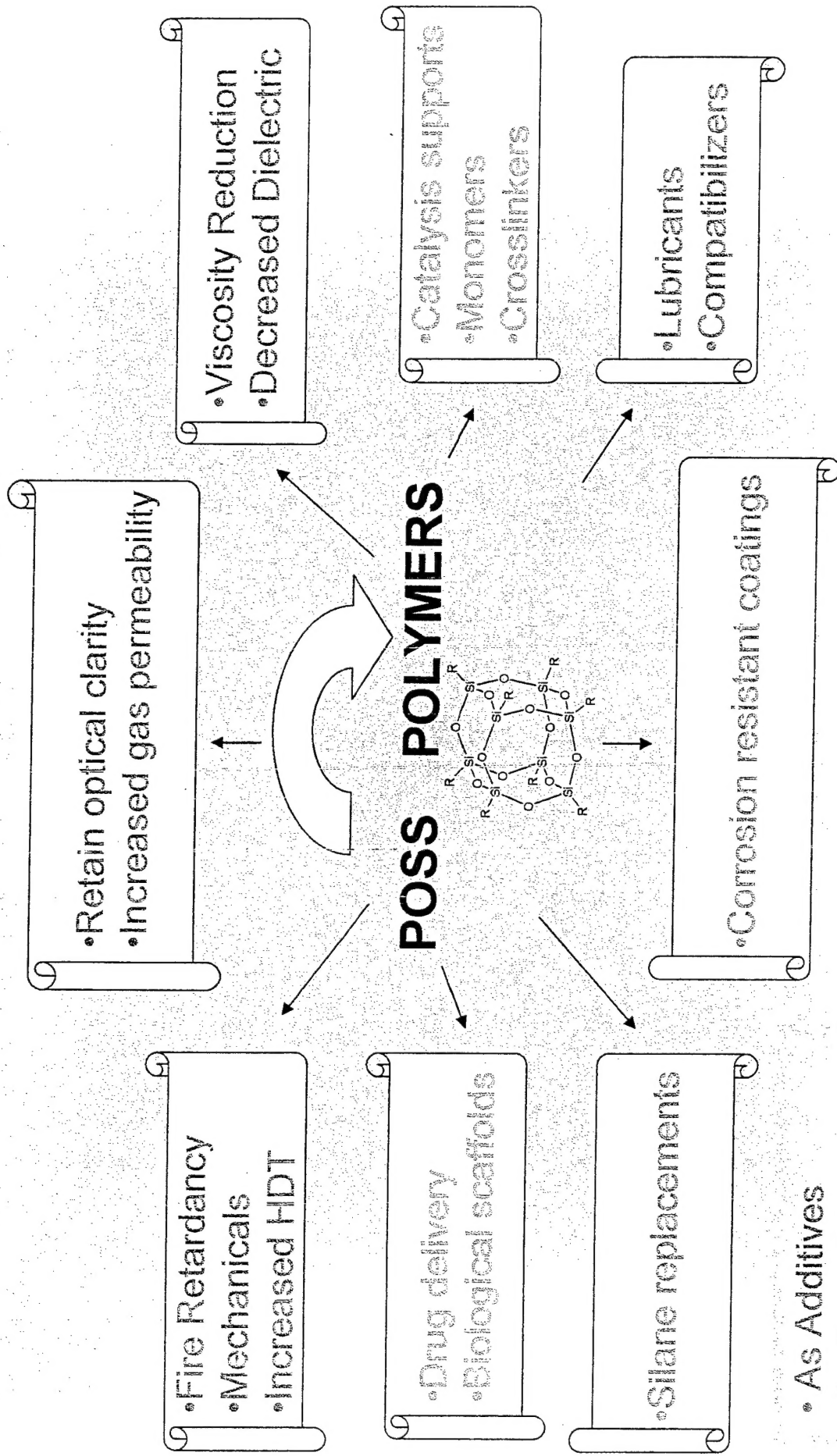
POSS – Polyhedral Oligomeric Silsesquioxanes



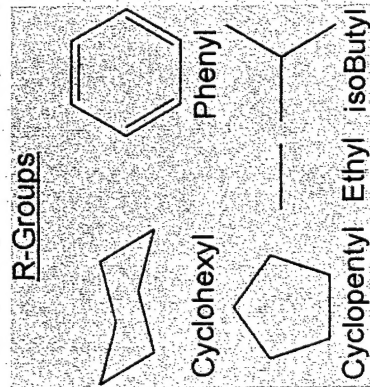
• Hybrid Plastics bridge the differences between ceramics and polymers

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# Introduction to POSS

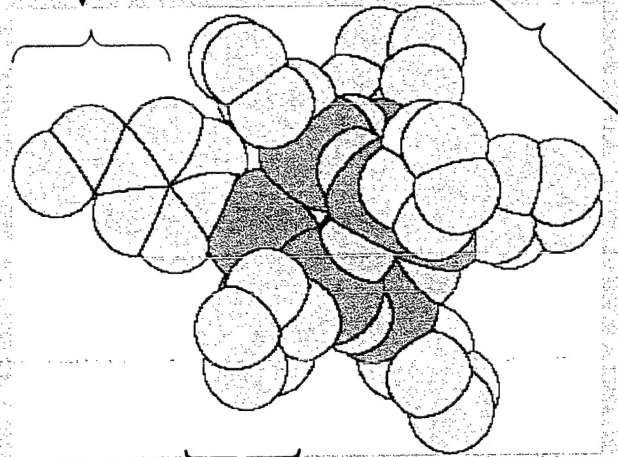


# Anatomy of a POSS Nanostructure



Nonreactive organic (R) groups for solubilization and compatibilization.

Nanoscale in size with an Si-Si distance of 0.5 nm and a R-R distance of 1.5 nm.



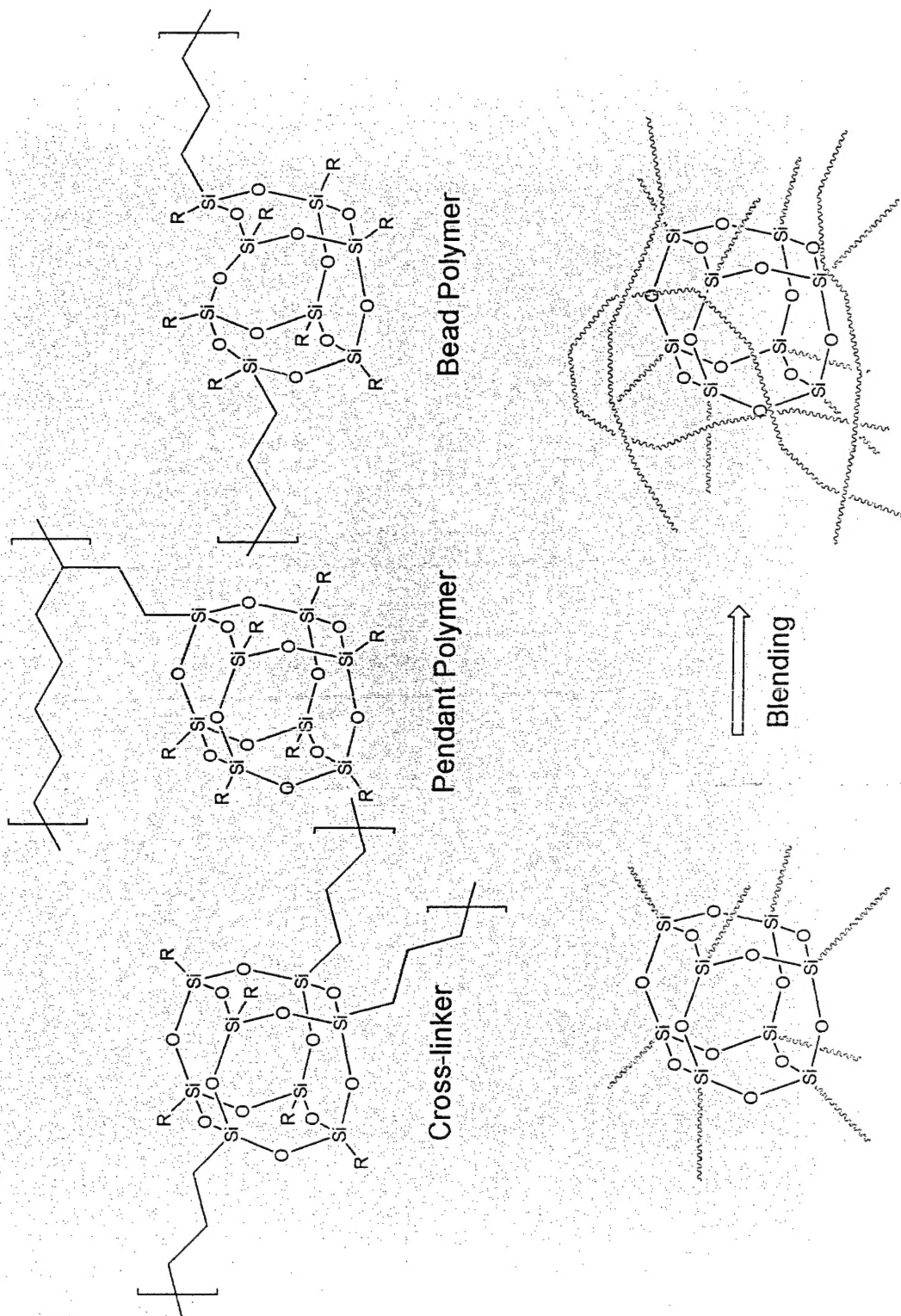
May possess one or more Functional groups suitable for Polymerization or grafting.

Thermally and chemically robust hybrid (organic-inorganic) framework.

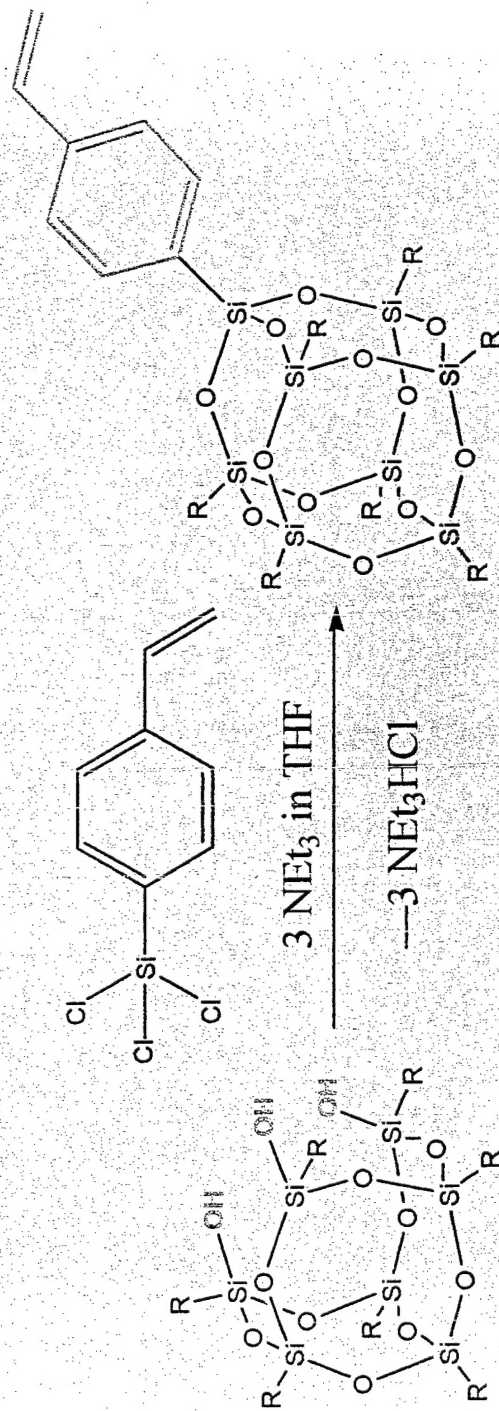
Precise three-dimensional structure for molecular level reinforcement of polymer segments and coils.



# POSS Polymer Incorporation

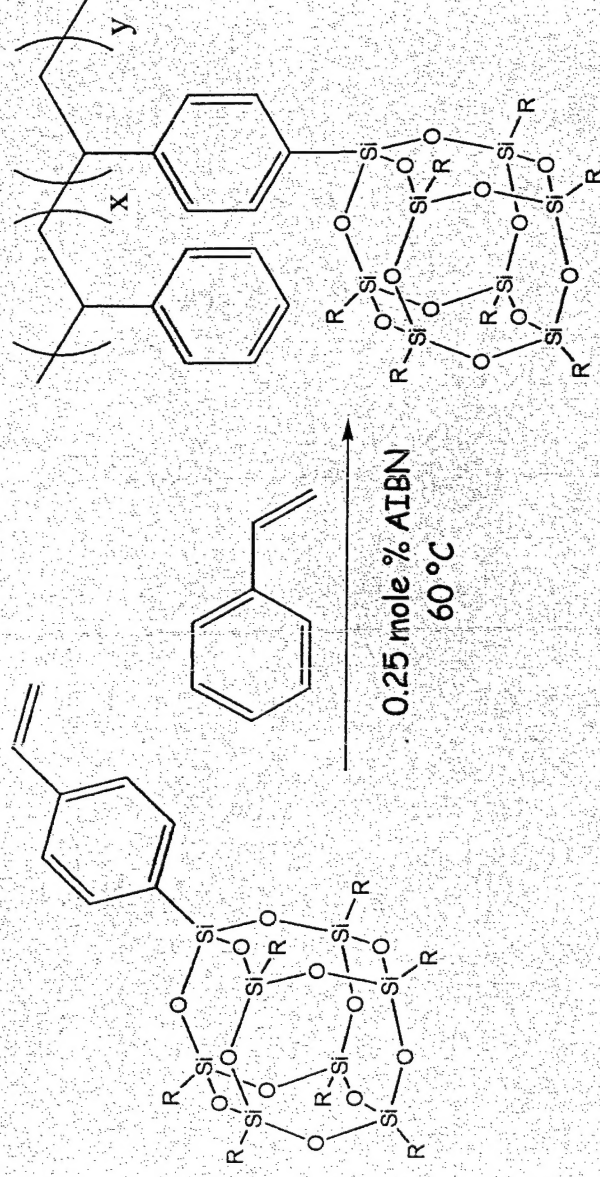


# POSS Styrene Monomer Synthesis



- High-yield syntheses
- Phenyl derivative requires inverse addition
- J. Inorg. Organomet. Polym., Vol 11, 2002, p. 155

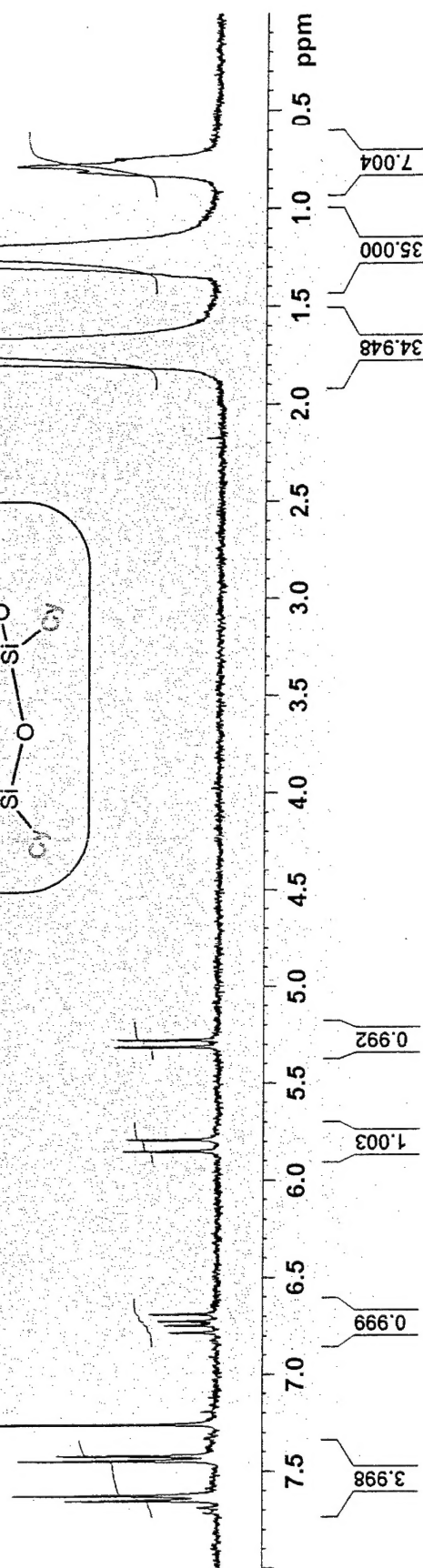
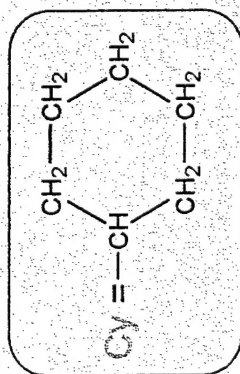
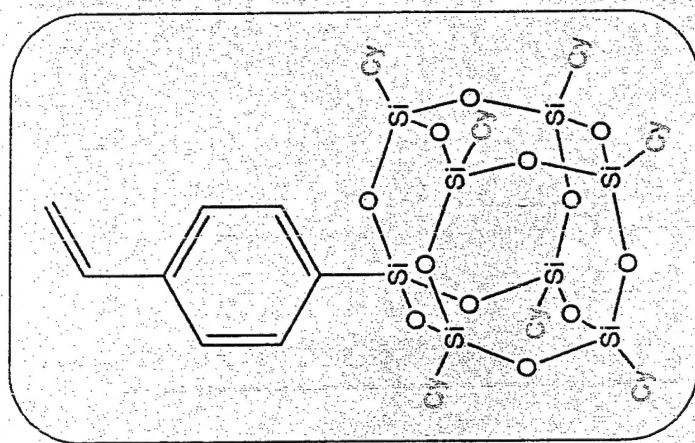
# POSS-Styrene Copolymer Synthesis



- Solution polymerization in toluene or bulk polymerization possible
- Polymerization is limited by solubility of the POSS-macromer
- Isobutyl-POSS is the most soluble, Phenyl-POSS the least soluble
- Macromolecules Vol. 29, 1996 p. 7302



# **<sup>1</sup>H NMR Cyclohexyl POSS monomer**



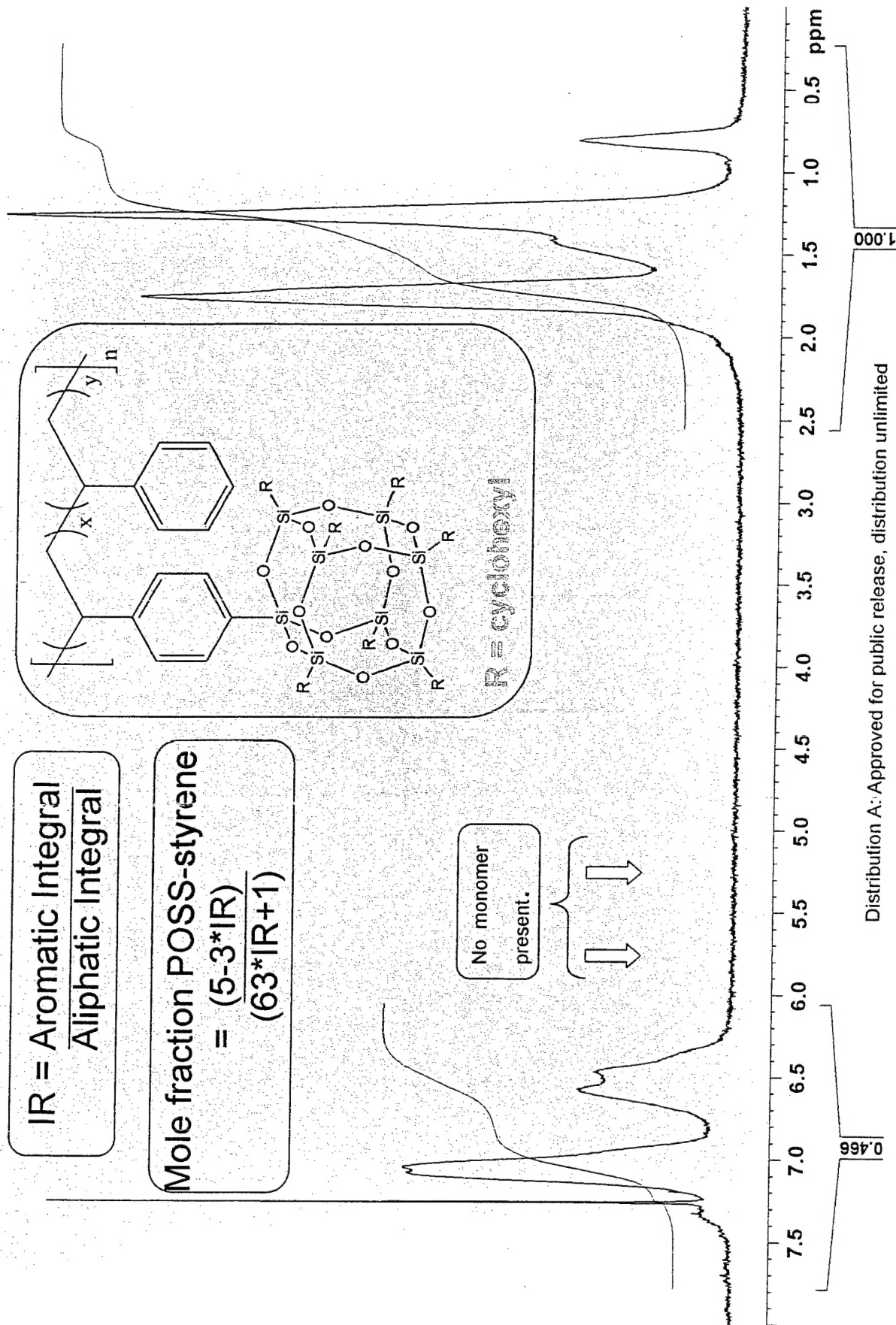
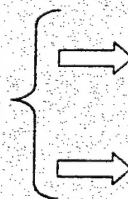
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# <sup>1</sup>H NMR 50 wt % CyPOSS-PS copolymer

IR = Aromatic Integral  
Aliphatic Integral

Mole fraction POSS-styrene  
=  $\frac{(5-3*IR)}{(63*IR+1)}$

No monomer  
present.



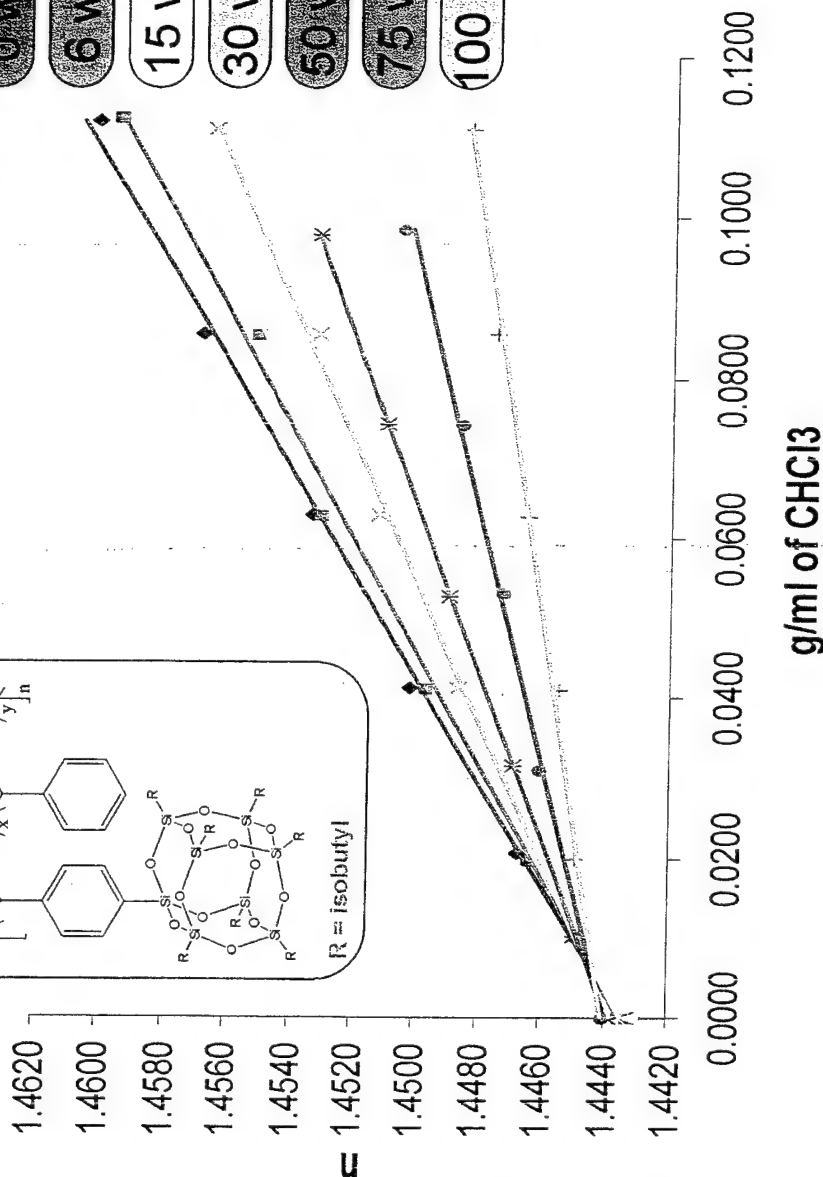
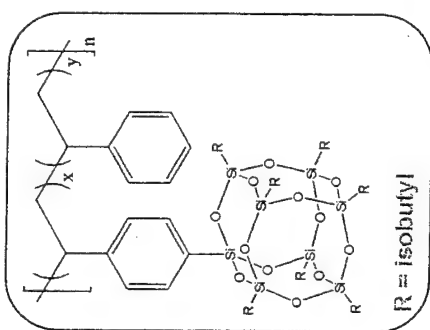
# The Specific Refractive Index Increment and Poss Polymer

- The Specific Refractive Index Increment,  $dn/dc = (n - n_0)/c$ , is the change in RI with change in concentration.
- It is a constant value for a dilute polymer in solution at constant temperature, pressure and wavelength.
- It is useful for determining the  $M_w$  of a polymer by light scattering (GPC).
- For copolymers composed of two monomers, the  $dn/dc$  is an additive function of the individual weight fractions.

# WAY FIND THE dn/dc VALUES?

- To accurately determine the weight average molecular weights of various POSS-polymers.
- To generically parameterize each POSS type (R = cyclohexyl, isobutyl, phenyl etc.) in order to predict POSS-polymer dn/dc values.
- To provide a quick and accurate method to determine POSS % incorporation in any polymer system.

# RI vs Concentration of BIPSS-PS copolymer



wt.% POSS-Styrene

Slope = dn/dc

$$Y = 0.1327x + 1.4434$$

$$R^2 = 0.9966$$

$$Y = 0.1428x + 1.4435$$

$$R^2 = 0.9958$$

$$Y = 0.1327x + 1.4436$$

$$R^2 = 0.9929$$

$$Y = 0.1159x + 1.4437$$

$$R^2 = 0.9976$$

$$Y = 0.0954x + 1.4439$$

$$R^2 = 0.9999$$

$$Y = 0.0636x + 1.444$$

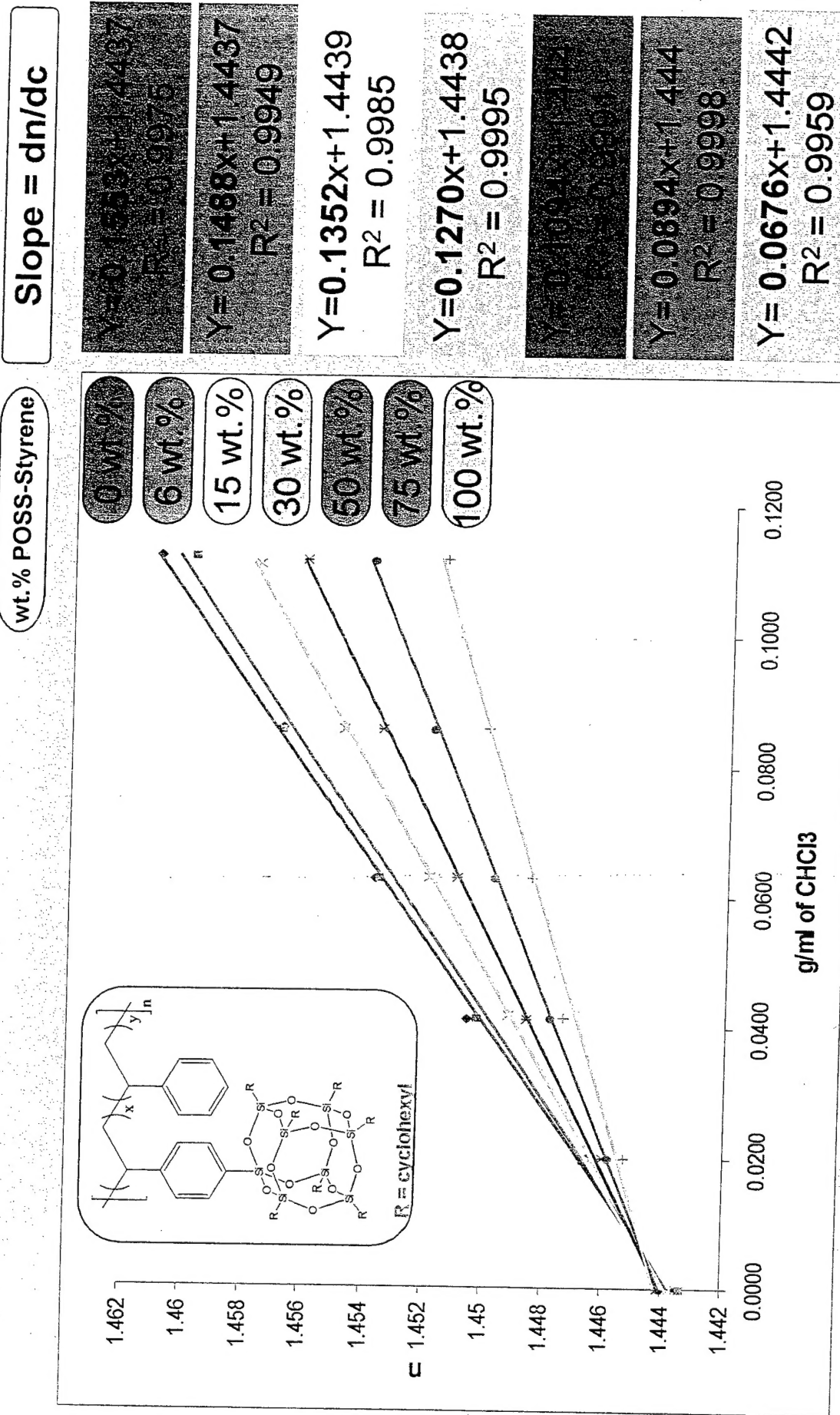
$$R^2 = 0.9962$$

$$Y = 0.0402x + 1.444$$

$$R^2 = 0.9976$$

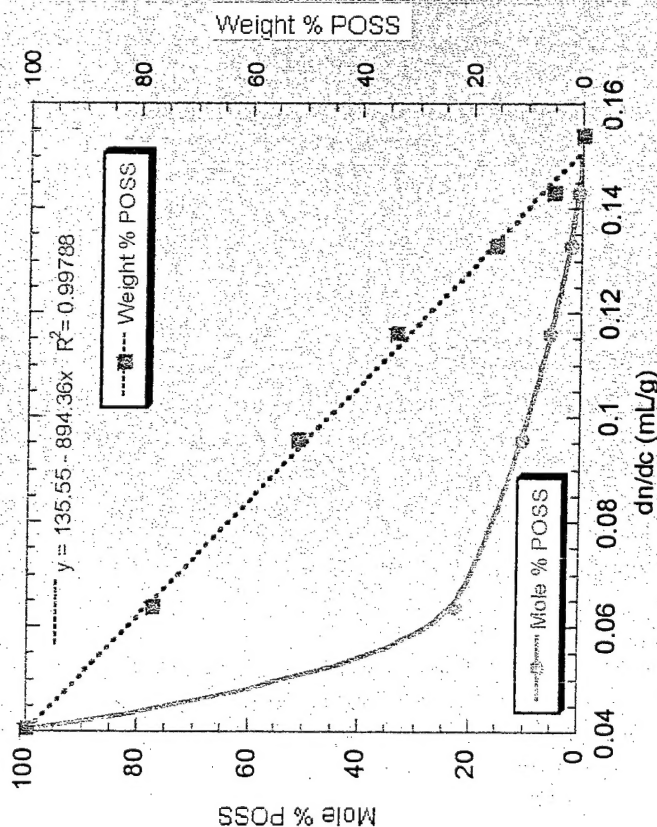


# R1 vs. Concentration of CyPSS-co-polymer

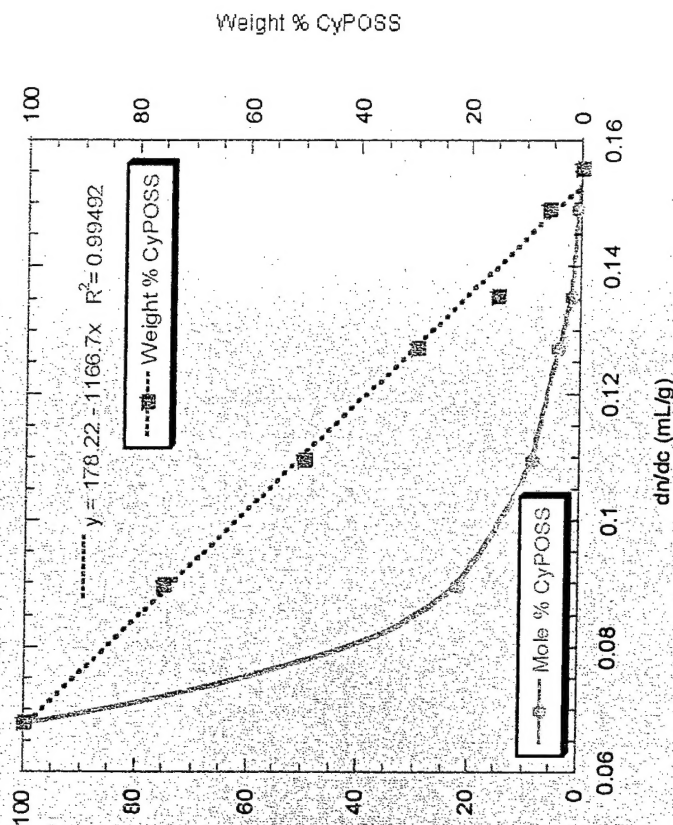


# dn/dc vs wt% and mole%

iBuPOSS-PS copolymer



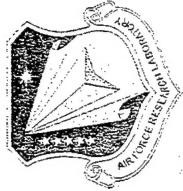
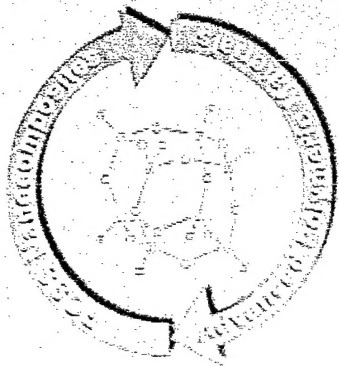
CyPOSS-PS copolymer



- Clearly, dn/dc is linear with respect to weight % POSS not mole % POSS; changes in refractive index are proportional to the volume occupied by the polymeric components. A typical POSS monomer is about 10X more massive than a typical organic monomer.
- Note that the dn/dc value decreases with increasing POSS incorporation.

## Summary & Future Work

- There is a linear relationship between weight % POSS and the  $dn/dc$  of a styrene copolymer.
- To graph a  $dn/dc$  / weight % POSS relationship for any new POSS polymer it is reasonable to just measure the  $dn/dc$  values of the 0 and 100 % POSS polymer.
- We intend to prove this concept for other glassy (Acrylics) and rubbery (Norbornenes) POSS copolymers.



# **The POSS-Polymer Working Group Air Force Research Lab - Edwards**

- |                      |                           |
|----------------------|---------------------------|
| • Dr. Shawn Phillips | • Maj Constance Schlaefer |
| • Mr. Patrick Ruth   | • Mr. Brian Moore         |
| • 2 Lt Will Cooper   | • Dr. Sandra Tomczak      |
| • Mr. Scott Barker   | • 2 Lt Amy Palecek        |
| • Dr. Rusty Blanski  | • Dr. Timothy Haddad      |
| • Dr. Joseph Mabry   | • 2 Lt Laura Moody        |
| • Mrs. Sherly Largo  | • Dr. Darrell Marchant    |

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